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Subject Details: Chemistry SL Paper 3 Markscheme Mark Allocation Candidates are required to answer questions from TWO of the options [2 20 marks]. Maximum total = [40 marks]. 1. A markscheme often has more marking points than the total allows. This is intentional. Do not award more than the maximum marks allowed for part of a question. 2.

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3H<sub>2</sub>O and coefficient of 3 on stearic acid; [3] Accept displayed or condensed formulas for molecules. (b) (i) number of grams/amount/mass of iodine/I<sub>2</sub> that add to/react with 100 g of

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1,3,5-hexatriene has three conjugated double bonds, so least energy; 1,3-pentadiene has two conjugated double bonds, it needs more energy; Accept alternative wording for M2 and M3: 1,3-pentadiene has two conjugated double bonds, 1,3,5-hexatriene has three / conjugation increases from 1,3-pentadiene to 1,3,5-hexatriene; thus need more energy;

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(b) Paper chromatography: water (in the fibres of the paper); Do not accept just paper. Column chromatography: alumina/Al<sub>2</sub>O<sub>3</sub>/aluminium oxide / silica (gel)/SiO<sub>2</sub>/silicon dioxide; [2] A4. Explanation of colour: [3 max] d orbitals splits (into two levels); due to repulsion between d electrons and non-bonding electrons on ligand / due to

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